## Equations and Stuff (Week One)

PV = nRT

$$E_{photon} = hv = \frac{hc}{/}$$

$$t = \frac{stock}{flux} = \frac{1}{Sk}$$

$$P_T = SC_iP_i$$

$$P^o = \frac{M_{atm}g}{4\rho r^2}$$

**Derived Values** 

$$\overline{M}_{air}$$
 = 28.96 g/mol (tropo and strato)  
 $n_{air}$  = 2.69x10<sup>19</sup> molecules/cm<sup>3</sup> (at STP)

Useful Bits

STP = 273K, 101,300 Pa (O°C, 1.00 atm)

Planck's constant; h=6.626 x  $10^{-34}$  J s

Speed of light in vac;  $c = 3.00 \times 10^8 \text{ m/s}$ 

Acceleration due to gravity;  $g = 9.8 \text{ m/s}^2$ 

Universal gas constant; R = 0.0206 L atm mol<sup>-1</sup> K<sup>-1</sup> = 8.314 J mol<sup>-1</sup> K<sup>-1</sup>

Pa = N m<sup>-2</sup> N = kg m s<sup>-2</sup> J = N m = kg m<sup>2</sup> s<sup>-2</sup>

## Terminology (Week ONE)

Source, Sink

Reservoir/Compartment

Transformation

Thermodynamics

Kinetics

Residence Time/Lifetime

**Primary Pollutant** 

Secondary Pollutant

Box Model

Troposphere, Stratosphere, Mesosphere, Thermosphere

Radical

Covalent oxide

CHEM 302: Atmospheric Environmental Chemistry